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INVESTIGATION OF THE STRUCTURE AND PHYSICOCHEMICAL PROPERTIES OF ALLOYS OF MOLYBDENUM DISILICIDE WITH NICKEL, COBALT, VANADIUM AND NIOBIUM

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## INVESTIGATION OF THE STRUCTURE AND PHYSICOCHEMICAL PROPERTIES OF ALLOYS OF MOLYBDENUM DISLLICIDE WITH NICKEL, COBALT, VANADIUM AND NIOBIUM

#### -USSR-

[Following is a translation of the article "Issledovaniye stroyeniya i fiziko-mekhanicheskikh svoystv splavov disilitsida molibdena s nikelem, kobaltom, vanadiyem i niobiyem" by Ye. M. Savitskiy and V.V. Baron in Doklady Instituta Metallurgii imeni A.A. Baykova (Works of the Institute of Mettalurgy imeni A.A. Baykova), No 5, Production Metallurgy, Physical Metallurgy, and Physicochemical Methods of Research, Moscow, 1960, pages 156-161.]

Investigations of the phase diagrams of the alloys of certain refractory metals with silicon (1-4) show that the silicides formed in the systems are likewise very refractory and many of them, unlike the metals forming them, possess chemical resistance to oxidation at high temperatures. One of these compounds is molybdenum dicilicide MoSi<sub>2</sub>.

It is pointed out in the literature (5, 6) that this compound, which melts at 2030°, does not oxidize at 1650° and has great strength at high temperatures even in lengthy tests: with a 100-hour stress at 98° the strength of MoSi<sub>2</sub> is 21 kg/sq mm. Its electric resistance is close to that of such materials as carbon steel, which evidences the metallic nature of this compound (7). However, because of the high brittleness of the alloys composed of molybdenum dicilicide, especially at room temperature, its use as a heat-resisting material is difficult.

In recent years a number of papers have appeared in print on the manufacture of metal ceramics consisting of a mixture of molybdenum dicilicide with various metallic additions (Pt., Ni., Co., Ti., Zn., Hf and others) introduced for the purpose of increasing plasticity (8-11).

[As a result, metal-ceramic materials have been obtained]

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With good strength at high temperatures and with a certain plasticity, but possessing low heat resistance.

There are almost no data on the structure and properties of cast alloys consisting of a mixture of molybdenum dicklicide with other elements, which is apparently due to the difficulties of obtaining such samples (refractoriness, oxidability of the component elements of the alloys, great . hardness, brittleness, etc.).

The present work is devoted to a study of the properties and structure of alloys of molybdenum dicilicide with nickel, cobalt, vanadium and niobium. For this purpose we prepared alloys of the quasibinary sections MoSi<sub>2</sub>-Ni, MoSi<sub>2</sub>-Co, MoSi<sub>2</sub>-V, and MoSi<sub>2</sub>-Nb. The alloys were prepared in three ways: melting in an arc furnace in an argon or helium atmosphere with an infusible tungsten electrode; melting in a high-frequency vacuum furnace; and production of alloys by remelting the molybdenum dicilicide prepared in the arc furnace in an open high-frequency circuit with argon blast and subsequent introduction of the above-mentioned metals into the alloys. The composition of the cast alloys by charge and chemical analysis is given in table 1.

The alloys with niobium and vanadium were prepared in the arc furnace. The alloys with nickel and cobalt were at first melted in the high-frequency vacuum furnace and poured into a copper mold, but because of the difficulty of obtaining non-porous high-grade ingots in this manner, owing to the high melting point, this method of pouring was replaced by another and the samples consisting of Mo and Si with Ni or Co were prepared in an open circuit with argon blast. The initial material was molybdenum dicilicide, previously melted in an arc furnace, to which mickel or cobalt was then added.

We used metals of the following purity: molybdenum 99.9%, silicon 99.8%, nickel 99.9%, cobalt 99.5%, vanadium 95.6%, niobium 99.3%. Samples were made of the alloys for measuring the hardness, tensile strength and plasticity and for investigating the microstructure, as well as for determining the melting point of the alloys by the drop method.

The strength and plasticity of the brittle and very hard samples were determined under uniaxial compression and, in the case of alloys amenable to cutting, their properties under stretching were also determined.

Study of the microstructure of the alloys with vanadium and niobium cast and roasted in a vacuum at 1200° for 6 hours showed that all of them are two-phased. The alloys contain-

О Состав по плихте, вес. %					(2)	Состав	no xem. Rei	алвэу, в	ec. %		
Mo	51	NI	Co	Nb		Мо	Si	Ni	Co	Nb	V
	1					(3)					
56,5	33,5	10				Осталь-	33,8	10,3		-	_
50,5	29,5	20				ное	29,02	19,1	_		
44	26	30				,	23	28,75			
38 '	22	40					16	36,35		_	
31,5	18,5	50	-		-	*	15,1	51,04			
25,5	14,5	60					12,7	57,66			
25	14	61				<b>3</b>	8,8	61,28			-
22	13	65		******			13,21	64,99			
20,9	12,1	67				*	11,94	66,56		****	
20,2	11,8	68	3			,	12,1	66,1			
19	11	70					10,8	64,49			
13	7	80					7,38	80,12			
6,5	3,5	90			where	35	3,37	90,63	gi	<u> </u>	gast-w
56,5	33,5	10				39	31,74		10,02		and the
50,5	29,5	20				10	27,7	glamosti.	20,4		
44	26	Chapterings.	30	approx.		,	24,6	-	39,9		) and the same of
38	22	40.00	40	aparen.		*	22,4		40,1	-	-
31,5	18,5	yeste	50	-			18,59	-	49,78		-
25,5	14,5	*****	60				14,98	N/Sideral S	59,9		arma)
19	11,0		70	-			10,95	district*	69,4	,	nice and a significant party.
<b>1</b> 3	7,0	nije <del>od</del> i/	80			16	7,08		80,04		guna
6,5	3,5	states and	90				3,56	*****	89,59		
56,5	33,5			10			32,0	-		10,46	-
50,5	29,5	Halistanda		20	_		28,5			20,38	
44	26	<b>PR</b> ajes*		30		96	27,5		genos.	30,8	-
38	22			40	-		22,5	demains	-	39,34	****

Legend to Table 1: 1) Composition by charge, \$ by weight; 2) Composition by chemical analysis, \$ by weight; and 3) Residual.

[Table 1 continued on following page.]

				Table 1	(Conti	nued)	•	
1								
1		19	18,2		-	50.32		
E		,,			atreture			
70								
80			1					
90	_	1 -		1			_	
	10	•				31,0	0.00	
		•				_		
		*			- January			
	30	,	25,4				30,03	
	i	60 — 70 — 80 — 90 — 10 — 20 — 30	60 — 80 — 90 — 10 — 20 — 30	50     —	50     —      18,2     —       60     —      14,7     —       70     —      9,9     —       80     —      4,5     —       90     —      2,2     —       —     10      31,09     —       —     20      29,82     —       —     30      25,4     —	50     —     .     18,2     —     —       60     —     .     14,7     —     —       70     —     .     9,9     —     —       80     —     .     4,5     —     —       90     —     .     2,2     —     —       —     10     .     31,09     —     —       —     20     .     29,82     —     —       —     30     ,     25,4     —     —	50     —     .     18,2     —     —     50,32       60     —     .     14,7     —     —     60,08       70     —     .     9,9     —     —     70,25       80     —     .     4,5     —     —     81,5       90     —     .     2,2     —     —     91,0       —     20     .     29,82     —     —     —       —     30     ,     25,4     —     —     —	50     —     .     18,2     —     —     50,32     —       60     —     .     14,7     —     —     60,08     —       70     —     .     9,9     —     —     70,25     —       80     —     .     4,5     —     —     81,5     —       90     —     .     2,2     —     91,0     —       —     10     .     31,09     —     —     9,88       —     20     .     29,82     —     —     9,88       —     30     ,     25,4     —     —     30,03

21,27

19,04

14,27

10,78

7,38

40,21

47,31

59,88

70.5

80.68

90.35

ing from 40 50 60% V have small quantities of inclusions of the second phase and apparently in these areas of concentration a metal compound is formed in this system, as indicated by the increased hardness and higher-melting point (about 1950°) as compared with the other alloys of this system. No solubility was observed in the solid state of the alloys in the intervals of concentration investigated.

40

50

60

70

80

90

31,5

25,5

6,5

56,5

50,5

44,0

31,5

25,5

19

13

6,5

38

19

19

18,5

14.5

7,0

3,5

33,5

29,5

26

22

18,5

14,5

11.0

7,0

3,5

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The investigation of the microstructure of the alloys with nickel and cobalt case and roasted in air for 6 hours at 1200° revealed that solid solutions with a nickel or cobalt base are formed in these systems, and further increase in the MoSi2 content results in the appearance of eutectic. The melting point of the latter in the MoSi2-Ni system is about 1250°, and in the MoSi2Co system about 1300°. The other alloys also consist of two and three phases.

The hardness of the alloys was measured at 20 and 1000°.

This was done by the method of pressing in a pobedite [allow of tungsten carbide particles cemented with cobalt] cone with loads of 50 and 100 kg. In heating to 1000°, the alloys of molybdenum with silicon and vanadium and of molybdenum with silicon and niobium were blasted with argon to protect them from oxidation. The alloys of the MoSi.-Mi section proved very hard at room temperature, especiálly with a 50% Ni, 18.5% Si and 31.5% Mo content. The hardness of this alloy reached 700 kg/sq mm. With increase of the nickel content the hardness drops and with an alloy having 90% Ni it reaches 170 kg/sq mm. At 1000° the hardness values are much less; the greatest hardness in this case is acquired by alloys containing small quantities of nickel (for exsmyle 10%) and constructed mainly of MoSi2. An increase in the nickel content to 50% lowers the hardhess of these alloys at 1000°. When its concentration is from 50 to 90%, the bardness of the alloys changes little and averages 50 kg/sq mm (Table 2).

				Ta	ible 2	
	Prop		MoSi	Ni Alloys	P. S	177
Conepisanue Ni. Bec. %	Тверность при 10° Н, ке/мм²	(3) Тверлость при 1000° И <sub>н</sub> . кг/мм²	Пречность при сжатии овсяк, кг/мм²	Прочность при растижении оджима	Пластичность при сжатив є, %	Пластви- ность при растиже- нии 8, %
A second section of the second				·		
10	350	310	16		0	
20	450	175	26	ggipanit-	0	-
30	560	90	26	1.35 x 1.000	0	
40	620	<b>7</b> 5	62	gament der	0	
50	<b>7</b> 30	50	112		0	Made 197
. 60	620	60	60	ggly-repres	0	- Annian
61	620	65	35		. 0	gg-m-Julyk
65	540	45	130		30	- Committee
67	590	55	135	application.	20	-
68	540	50	170	deland	45	Special Services
70	560	40	240	18	53	0
80	410	45	250	85	49	4
90	170	35	galak sah	55	61	31
100	80	greens-		30		80
Legend:	(1) Ni con % by t	ntent, (2 veight	Hardnes at 90% H <sub>K</sub> , kg, sq mm	]	lardness a .000° I <sub>K</sub> , kg/sq	

- (4) Strength under compression o'Bcomp: k/g sq mm
- (5) Strength under tension of kg/sq mm
- (6) Plasticity under compression E, %

(7) Plasticity under tension

Table 3

Properties of Mo--Si--Co alloys

and the second of the second o	[6]				(/)	171
Содержание	Твердесть прв 20° Н <sub>к</sub> . ке/мм <sup>2</sup>	Твердость при 1000° Н <sub>н</sub> , кг/мм²	Прочность при сжатви ов сж. ке/мм.	Прочность при растяжении о <sub>в</sub> , ке/мм	Пластич- ность при сжатии е. %	Пластичность- при растиже- нен 8, %
10	1000	125	0,5	aprinter.	0	
20	300	150	17		0	agishines.
30	<b>62</b> 0	440	19		. 0	
40	470	175	1,0		0	All Constitution of the Co
50	490	105	31		0	-
60	450	75	110		0	
70	450	50	215	46	20	0
80	430	40	225	98	28	1,5
90	350	39	190	66	30	2,1
100	200	30	55		32	

- (1) Co content % by weight
- (2) Hardness at 20°  $H_K$ , k/g sq mm
- (3) Hardness at 1000° HK, k/g sq mm

- (4) Strength under compression Bcomp: k/g sq mm
- tension σ<sub>B</sub>, kg/sq mm
- (5) Strength under (6) Plasticity under compression E, %

(7) Plasticity under tension 5. %

Table 4

Properties of Mo--Si--V alloys

Содержание	Твердость при 20°	Thephocit uph	Прочность при	Пластичность при сматив
10 20 30 40 50 60 70 80 90	940 830 620 610 590 1500 830 620 590 250	400 390 495 400 535 — 745 390 165	0,6 3 10 4 3 10 42 53 165 140	0 0 0 0 0 0 0 0 0 13,7

Legend:

(1) V content, % by weight

(2) Hardness at (3) Hardness at 20°  $H_K$ , 1000°  $H_K$ , k/g sq mm

(4) Strength under compression OBcompr. kg/sq rom

(5) Plasticity under compression  $\mathcal{E}$  , %

	And the second s	Properties of Mo-SiNb alloys						
Cogephanne Nb. sec. %	(2) Тверцость при 20° И <sub>н</sub> , ка/мм <sup>3</sup>	Твердость при 1000° Н <sub>н</sub> , ке/мм <sup>2</sup>	Прочность при сматии ов см' ка/мма	Плястичность при сматия				
	720	_	40	0				
10 20	780		12	0				
30	880		0	0				
40	780		70	0				
50	490		18	0				
60	830		30	0				
70	590		55	0				
80	590	_	205	4				
90	390	310	185	15				
100	150	70	· Distribution	41				

Legend:

- (1) Nb content, % by weight
- 20° H<sub>K</sub>, k/g sq mm
- (2) Hardness at (3) Hardness at 1000° H<sub>K</sub>, kg/sq mm
- (4) Strength under compression O Bcompr. kg/sq mm
- (5) Plasticity under compression

The strength of alloys with small nickel contents is small and they are brittle. With increase in its content, the strength grows, and alloys with 70 and 80% Ni are strongest (240-250 kg/sq mm). An allow with 90% Ni does not shatter under compression.

The plasticity of alloys grows with increase in nickel content and in an alloy with 80% Ni it is 49% under compression, and in an alloy with 90% Ni it is 62%.

In stretching tests the alloy with 80% Ni ( = 85 kg/ sq mm) proved the strongest and the alloy with 90% Ni the most plastic ( $\int = 31\%$ ).

A similar change in hardness at 20 and 1000° and of strength under compression and tension was also observed in calloys of the MoSi<sub>2</sub>-Co section with an increase in the cobalt content (Table 3), with the sole difference that the greatest hardness in this case was observed in the alloy with 30% Co. Alloys with 70-80% Co possess the greatest strength under compression (200-220 kg/sq mm). Under tension the strength of the alloy with 80% Co is greatest and amounts to 98 kg/sq mm. The plasticity of alloys with cobalt is less than that of alloys with nickel, and in the -alloy with 90% nickel it is 30% in the compression tests and 2.1% under tension.

The alloys of the MoSi<sub>2</sub>-V and MoSi<sub>2</sub>-Nb section proved very hard not only at room temperature, but also at 1000°. The alloys with vanadium are especially hard when they contain 60% of this metal. The hardness of the alloy with 50% V exceeds that of the pobedite cone at 20 and 1000°. It remains high when the vanadium content is 70-80% (Table 4). The alloys possess low strength under compression up to 70% V, but it grows with further increase of its content, and in the alloy with 90% V it is 165 kg/sq mm, which is greater than the strength of vanadium itself.

The MoSia-V samples oxidate strongly at high temperatures and especially with a high concentration of vanadium. Plasticity is detected only in the alloy containing 90% V and allowing of settling by 14% before shattering.

The hardness of the alloys of the MoSi2-Nb section is somewhat lower at room temperature than in the preceding case, but at 1000° the softening of the alloys is less and their hardness is so great that we did not succeed in measuring it with the pobedite cone (Table 5). One can only note that the alloy with 90% Nb at 1000° has a hardness of 310 kg/sq mm.

The strength of alloys with a small niobium content is also very low, especially in alloys with 30% Nb. When its content is raised to more than 30%, the strength grows and reaches a maximum at 80% NB (205 kg/sq mm). The strength of niobium under compression could not be determined because of its considerable plasticity.

The plasticity of the alloys of the MoSi2-Nb system is low and only when the Nb content is 90% is the alloy deformed before shattering under 15% compression. The oxidability of the alloys with niobium at 1000° is considerably less than when vanadium is added.

### CONCLUSIONS

- 1. The alloys of the MoSi<sub>2</sub>--Ni, MoSi<sub>2</sub>--Cc, MoSi<sub>2</sub>--V and MoSi<sub>2</sub>--Nb sections at room temperature possess high hardness and Tow strength and plasticity when they contain less than 60-70% by weight of the added elements. Hardness at 1000° is also high and reaches especially great values in alloys with niobium and vanadium (500-900 kg/sq mm). The strength-of the alloys grows when the content of metals added to the MoSi<sub>2</sub> is increased above 60-70%, and remains high in all alloys except those with vanadium, being 200-250 kg/sq mm.
- 2. The most promising in their mechanical properties and their resistance to oxidation at high temperatures are the alloys of the Mo-Si-Ni and Mo-Si-Co systems in the concentration interval from 70 to 80% Ni and from 70 to 90% Co. These alloys possess considerable strength, plasticity and comparatively high hardness at 1000°, and can be subjected to mechanical working.

The alloys with a high niobium content (80-90%) also possess considerable strength and very high hardness at 1000°, oxidizing moderately at high temperatures.

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